

by weight

Take 100g air = 80g N₂ + 20g O₂

deal with pure
STP

V ₁	STP O ₂ 20g
V ₂	STP N ₂ 80g

$$= \frac{80 \text{ mole } N_2}{28} + \frac{20 \text{ mole } O_2}{32}$$

$$V = \frac{80}{28} \cdot 22.4 + \frac{20}{32} \cdot 22.4 =$$

$$(2.857 + .625) \cdot 22.4 = 3.482 \cdot 22.4$$

$$\therefore V = 78.0 \text{ litres}$$

$$78000 \text{ ml} = A \left(\frac{80}{28} \cdot 22.4 + \frac{20}{32} \cdot 22.4 \right)$$

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$$A(40 + 10)$$

$$= 50A$$

One mole
of air
mixture

$$\frac{30}{18600}$$

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$$18 \text{ mol} = 1 \text{ mole}$$

$$\text{one mole of air} =$$

Notes of lecture
 in detail

$$\frac{100 \times 60 \times 22.4}{18}$$

$$= 14700 \text{ ml}$$

$$50A$$

3.2

1.4 / *initial mass*

sample of air = 22400 mil. $6.02 \times 10^{23} \times 19$
18 wt. $6.02 \times 10^{23} \times 10$

air = $194 \times 6.02 \times 10^{23}$
 $\frac{22400}{19 \times 6.02 \times 10^{23}}$ = 1.16×10^{-3}

H₂O

SCRIPPS

H₂O

862

air

from Page 731 EAW

$$\left[\frac{N}{\text{sec}} \right] = 1.75 \times 10^5 \left[\frac{\text{ft}^2 \text{ flux} \times \text{Med/sec/cm}^2}{\text{in}^2} \right]$$

$$= 1.45 \times 10^5 \left[\frac{\text{energy at rate / second}}{\text{in}^2} \right]$$

$$\left[\frac{\text{m}^2/\text{hr}}{\text{in}^2} \right] = (1000)(3600) \cdot 5 \times 10^{-5} \left[\frac{\text{energy at rate / second}}{\text{in}^2} \right]$$

$$= 52.3$$

Now it can be shown that the rates of the electrons of energy in liquid water to the electron energy in gas at 866 eV is about 866

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Imagine now a bubble of air absorbing from a hole in a wall flux energy at the rate of 1 med/sec/cm² or the equivalent rate 52.3 m²/hr.

A perfect "rometer" would read directly 52.3 scale units. The realising meter would read 52.3 e² = (52.3)(523) = 27.4 scale units of the meter was 7¹³ calibration.

Now let us imagine pouring water into the bubble while holding the bubble constant. The energy would now be available as potential energy in the electrons present. The water would be at about 866 med/sec/cm², or the equivalent of 866 x 52.3 m²/hr. However, the meter would still read 27.4 units in the scale m²/hr.

There are considerable apparent to actual dose rate units in the calibration of the meter. For 7¹³ calibration 1 m²/hr = 27.4 med/sec/cm².

Concentration of the substance of similar density, assumed

$$[\text{mass per cm}^3] = 1/30.17 = .0317 [\text{mils}^3/\text{cm}^3]$$

the latter being the opposite of mass equal on the
head of the probe.

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Now since we infer the absorption coefficient is
flux intensity even where is uniform and the activity
everywhere is uniform and. This seems only correct
if the activity density is numerically equal to
the rate of absorption per unit volume per unit time.

Thus we can put in this special case, the activity
density also,

$$[\text{mass per cm}^3] = .0317 [\text{mils}^3/\text{cm}^3]$$